

## Biosynthesis pathways

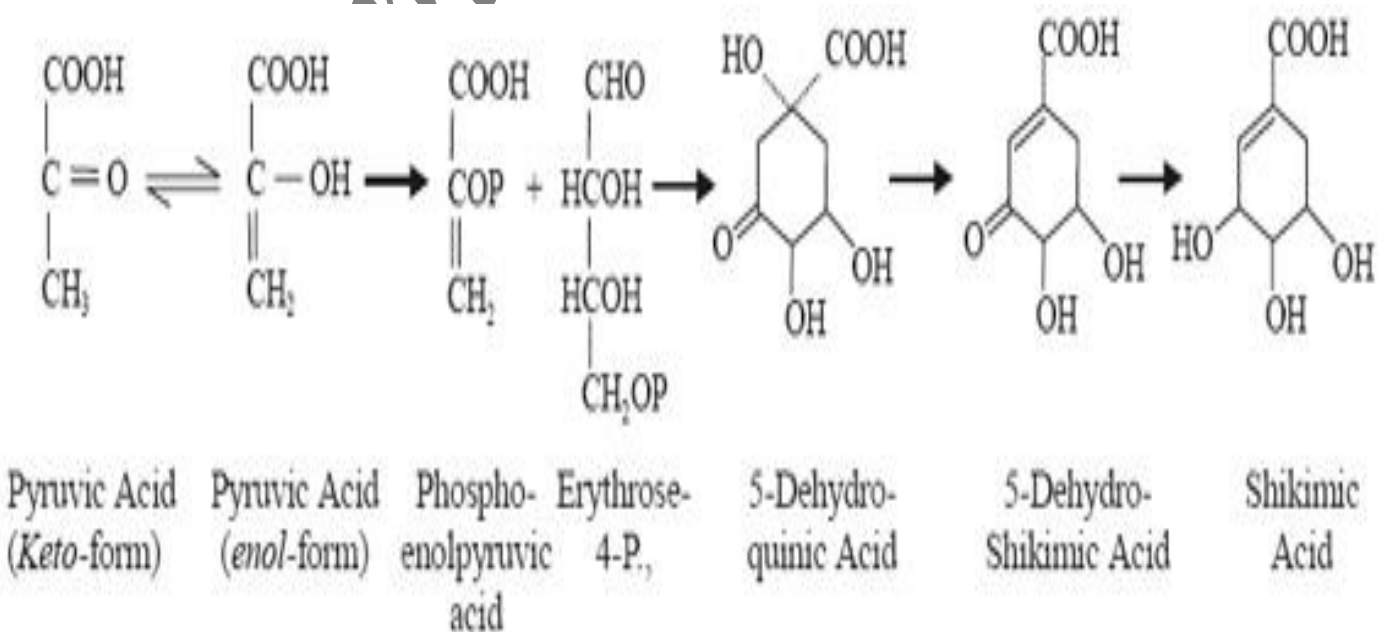
### 1. Biosynthesis of aromatic compounds: -

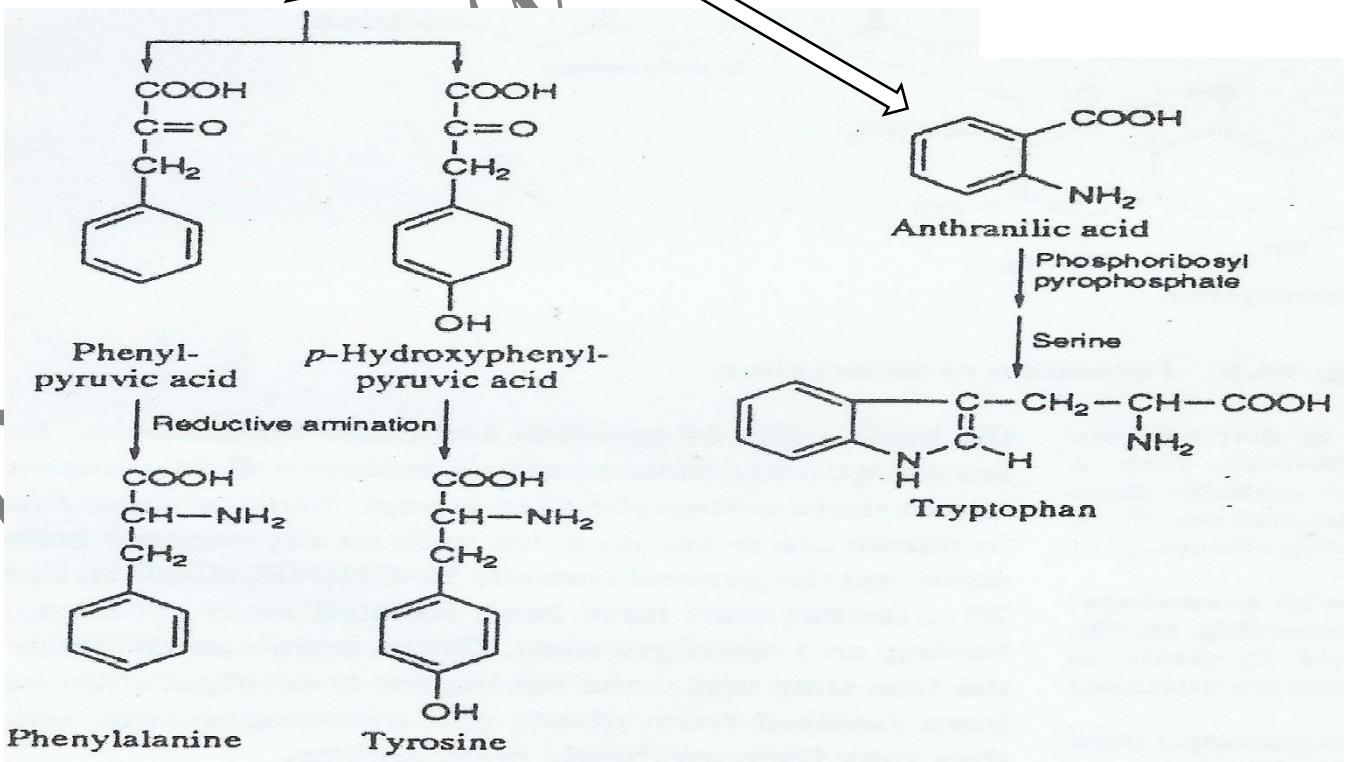
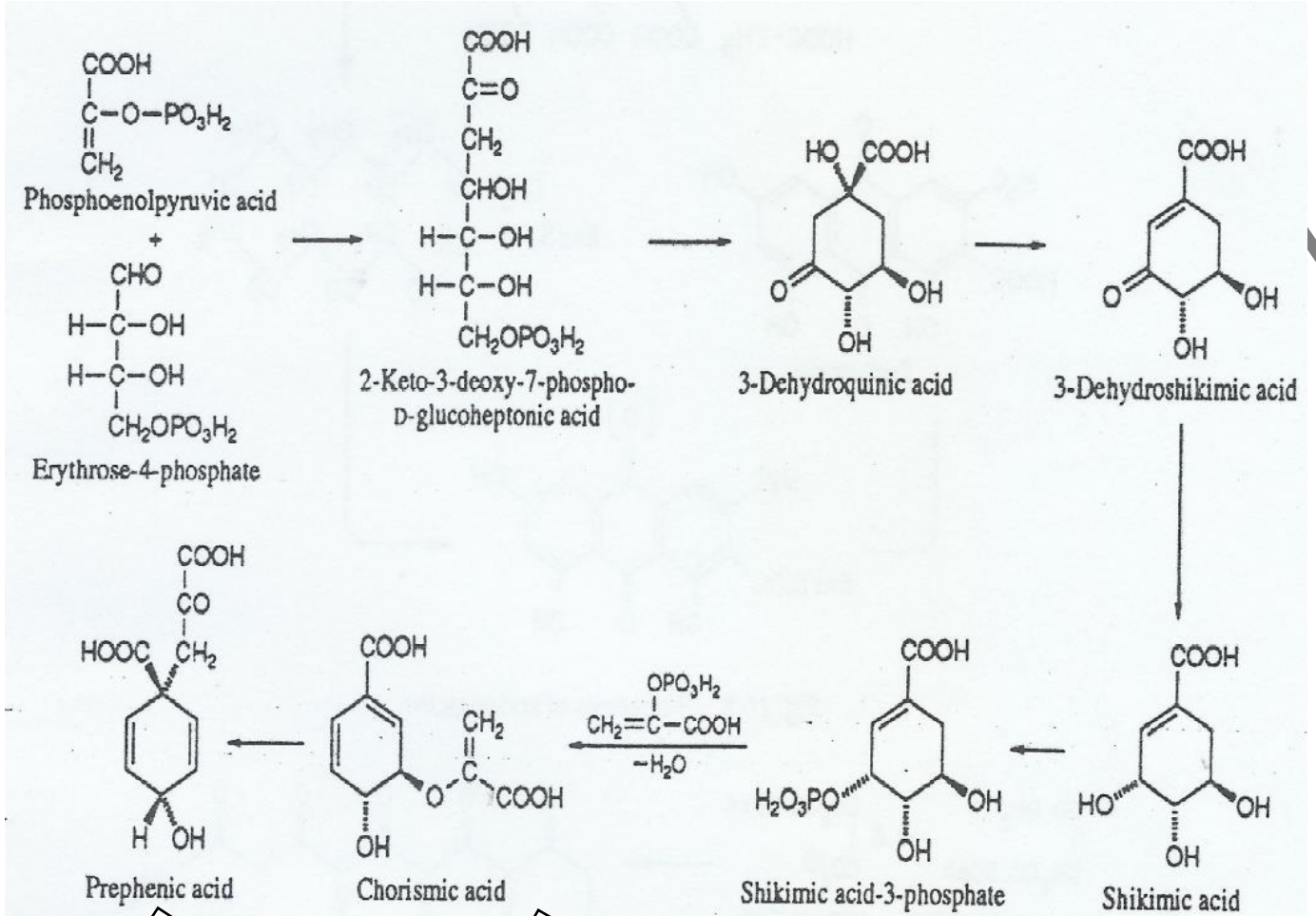
There are two pathways for the biosynthesis of aromatic compounds: -

**A. The shikimic acid pathway:** - Shikimic acid is biosynthesized from:

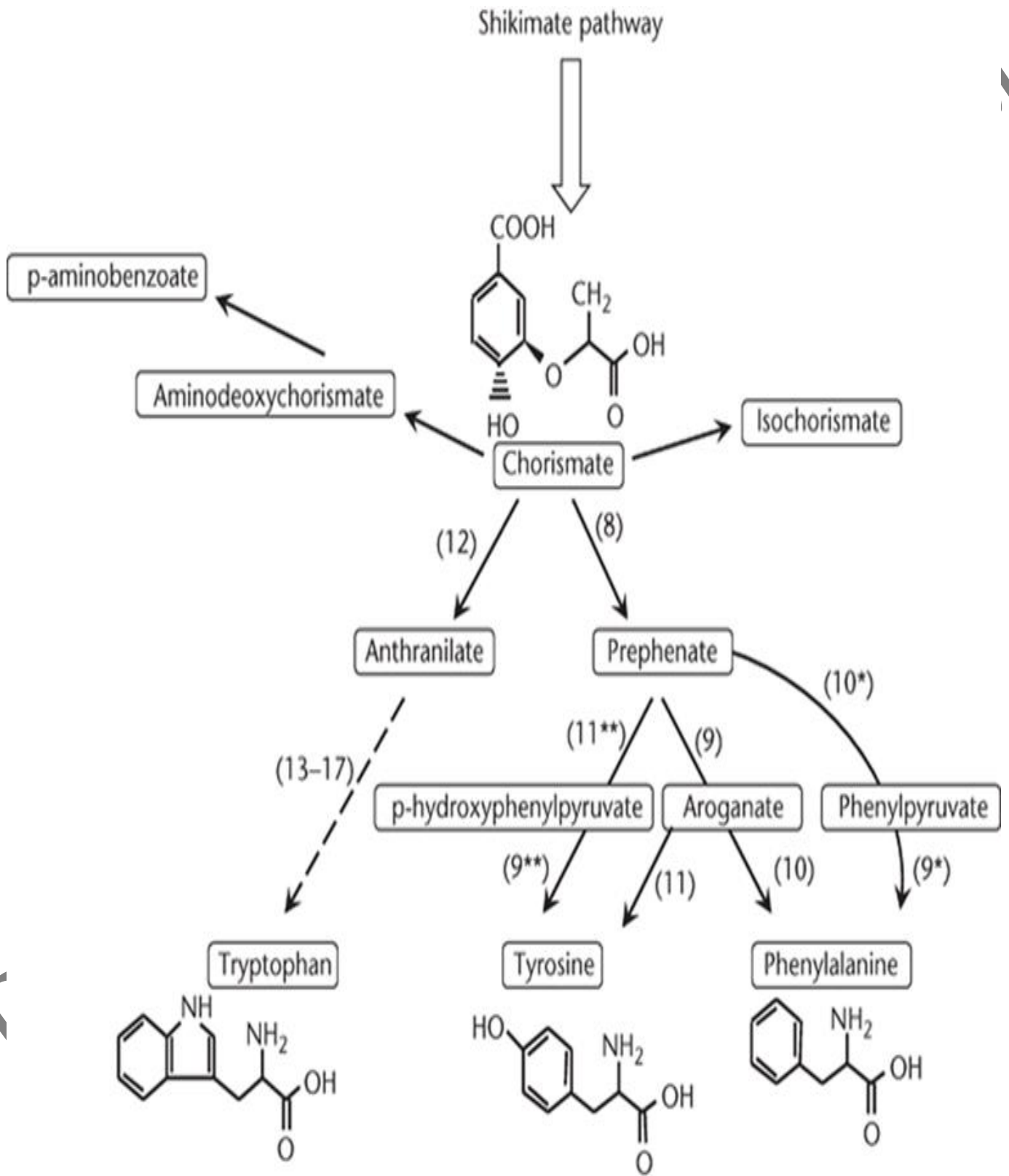
phosphoenol pyruvic acid + erythrose-4-phosphate

- + shikimic acid undergo a series of phosphorylated intermediates yield chorismic acid which is an important branch – point intermediate.
- + One branch leads two anthranilic acid then to tryptophan.
- + The other leads to prephenic acid which is the last non aromatic compound in the sequences.
- + Prephenic acid can be aromatized in two ways.
  - ✓ The first proceeds by dehydration and simultaneous decarboxylation to yield phenyl pyruvic acid, the direct precursor of phenyl alanine.
  - ✓ The second occurs by dehydrogenation and decarboxylation to give the P-hydroxy phenyl pyruvic acid, the precursor of tyrosine.

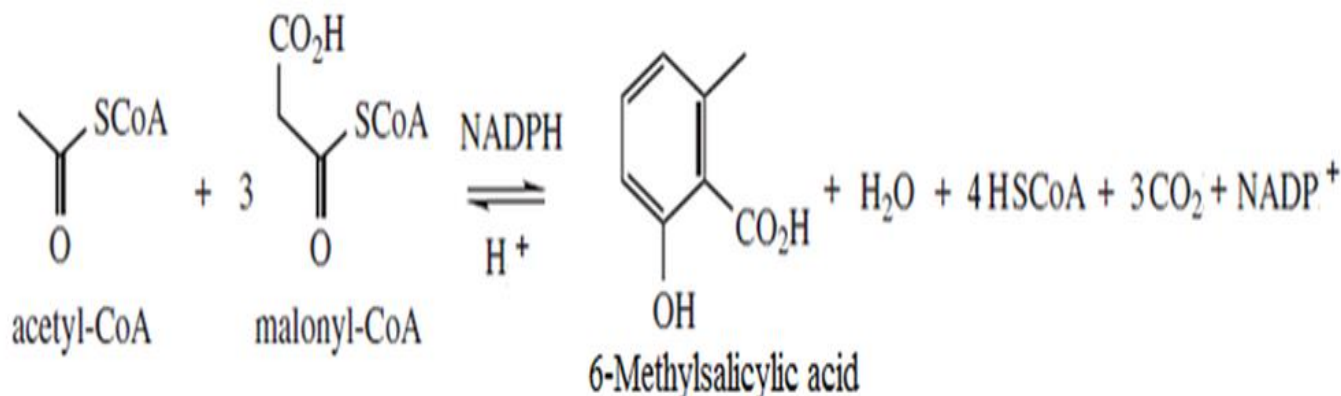




### Shikimate pathway for the biosynthesis of aromatic & phenolic compounds



**B. The acetate hypothesis:** - 4 acetate groups when linked together they give rise to the formation of 6-methyl salicylic acid. This was found when using labeled carbon in certain *Penicillium* species.



### ***Biosynthesis of non-aromatic compounds (acetate-mevalonate) pathway***

Three molecules of acetyl-coenzyme A are used to form mevalonic acid (MVA). Two molecules combine initially in a Claisen condensation to give acetoacetyl-CoA, & a third is incorporated via a stereo specific aldol addition giving the branched-chain ester  $\beta$ -hydroxy- $\beta$ -methyl glutaryl-CoA (HMG-CoA). The conversion of HMG-CoA into MVA involve a two-step reduction. The six-carbon compound MVA is transformed into the five-carbon phosphorylated isoprene units in a series of reactions involving decarboxylation/dehydration to give isopentenyl pyro phosphate (IPP).

C<sub>5</sub> isoprene units give rise a large structurally diverse family of natural products named terpenoids when these C<sub>5</sub> isoprene units joined together in a head-to-tail fashion. Typical structures contain carbon skeletons represented by (C<sub>5</sub>)<sub>n</sub>, & are classified as hemiterpenes (C<sub>5</sub>), monoterpenes (C<sub>10</sub>), sesquiterpenes (C<sub>15</sub>), diterpenes (C<sub>20</sub>), sesterterpenes (C<sub>25</sub>), triterpenes (C<sub>30</sub>) & tetraterpenes (C<sub>40</sub>).

Isoprene is produced naturally & the biochemically active isoprene units were identified as the diphosphate (pyrophosphate) esters **dimethyl allyl pyrophosphate (DMAPP)** & **isopentenyl pyrophosphate (IPP)**. Relatively few of the natural terpenoids conform exactly to the simple concept of a linear head-to-tail combination of isoprene are seen with geranyl (C<sub>10</sub>), farnesyl (C<sub>15</sub>), geranyl-geranyl (C<sub>20</sub>), squalene (C<sub>30</sub>), phytoene (C<sub>40</sub>).

## Mevalonate pathway

