

ENZYMES

- Enzymes are proteins that act as catalysts for metabolic reactions, making the reaction go faster.
- Enzymes work by lowering the energy of activation.
- Each enzyme is specific for a reaction.



• Enzymes are found in all living organisms and most cells contain hundreds of types which are constantly being manufactured and replaced.

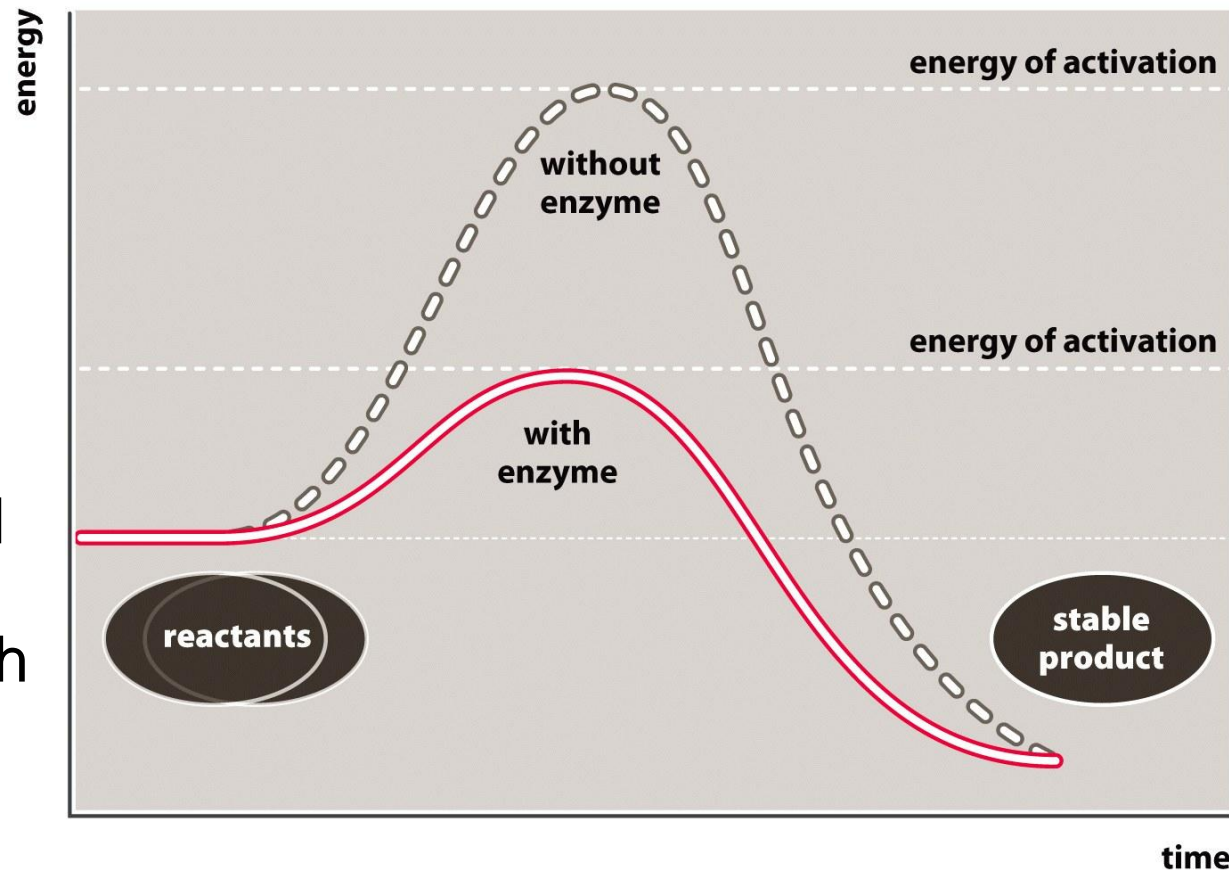


Figure 3.2 Microbiology: A Clinical Approach (© Garland Science)

PROPERTIES OF ENZYMES

- Enzymes are generally highly specific.
 - Some enzymes work with only one substrate, others work with a group of molecules.
 - Succinic dehydrogenase oxidizes only succinic acid
 - A proteolytic enzyme always degrades proteins because it reacts only with peptide bonds.
- The shape and electrical charges found at the active site allow for the reaction to work and are responsible for the enzyme's specificity.

PROPERTIES OF ENZYMES

- Enzymes have specific three dimensional shapes: if the shape changes, activity is inhibited.
- The shape of the molecule provides a importante site called the **active site**. It is here that:
 - The **substrate** fits into the enzyme's active site.
 - The enzyme and substrate interact to form the enzyme-substrate complex.
- The **active site** has to have the proper shape for the enzyme to work.

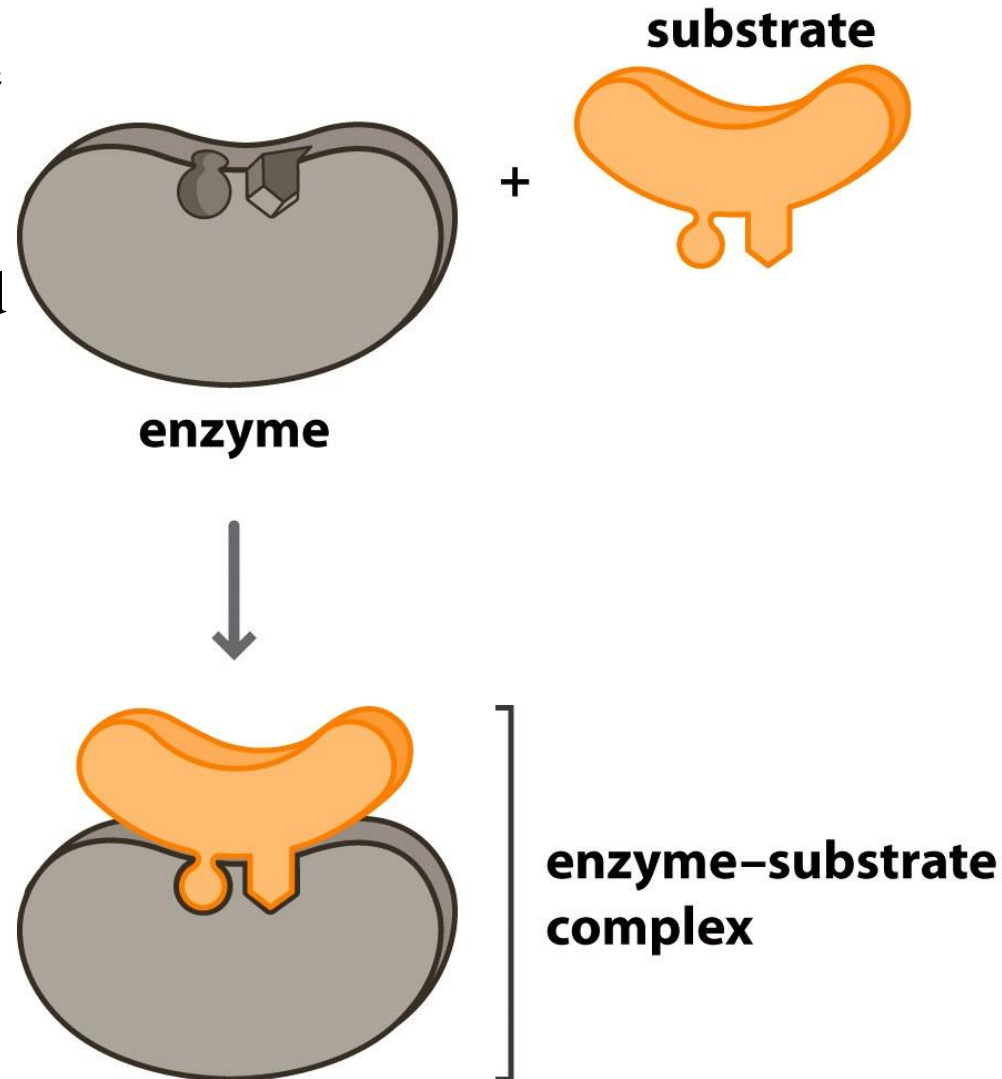


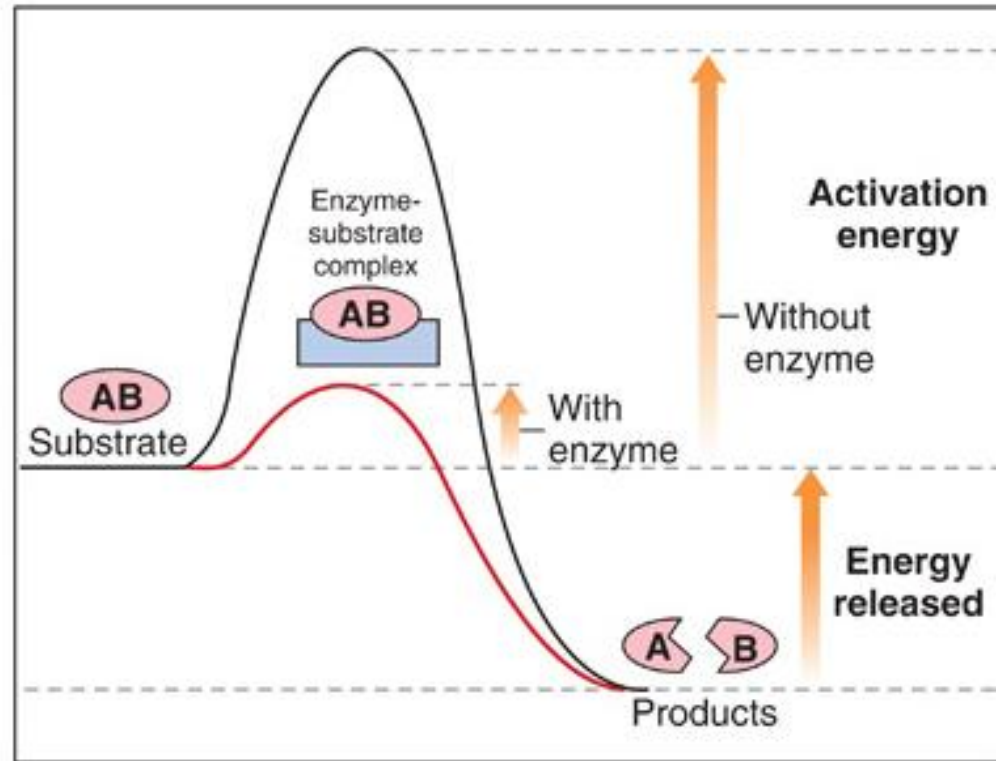
Figure 3.3 Microbiology: A Clinical Approach (© Garland Science)

Enzymes

- **Catalysts** are chemical substances that speed up a reaction without affecting the products.
- Catalysts are not used up or changed in any way during the reaction.
- **Enzymes** are important catalysts in living organisms.

Enzymes

- **Enzymes** reduce the amount of activation energy required for a reaction to proceed.
 - Enzymes are not used up or altered.
 - Products are not altered.
 - Energy released is the same.



COENZYMES AND CO-FACTORS

- Enzymes may be pure proteins or proteins plus **co-factors**
- **Co-factors** are helper substances that are inorganic ions such as magnesium, zinc, or manganese.
- **Coenzymes** are helper substances that are non-protein organic molecules.
- Co-factors or coenzymes bind to the active site and change the shape of the active site so the substrate now fits.

COENZYMES AND CO-FACTORS

- Coenzymes and co-factors can also be used as **carrier molecules**.
 - When a carrier molecule receives either electrons or hydrogen atoms, it becomes reduced.
 - When a carrier molecule releases electrons or hydrogen atoms, it becomes oxidized.

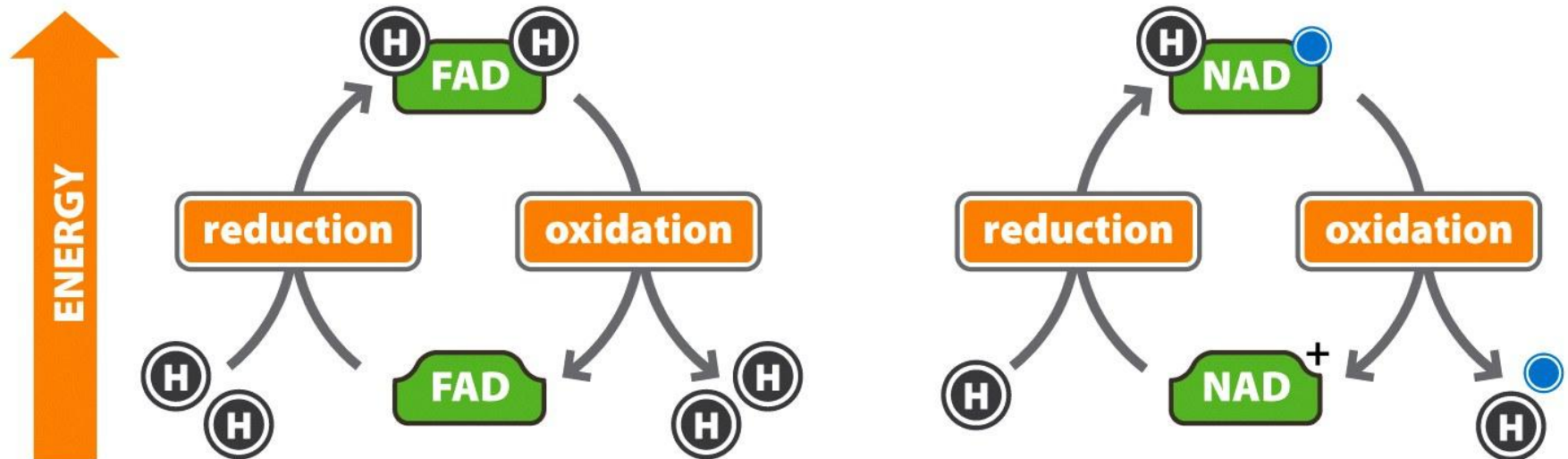


Figure 3.5 Microbiology: A Clinical Approach (© Garland Science)

COENZYMES AND CO-FACTORS

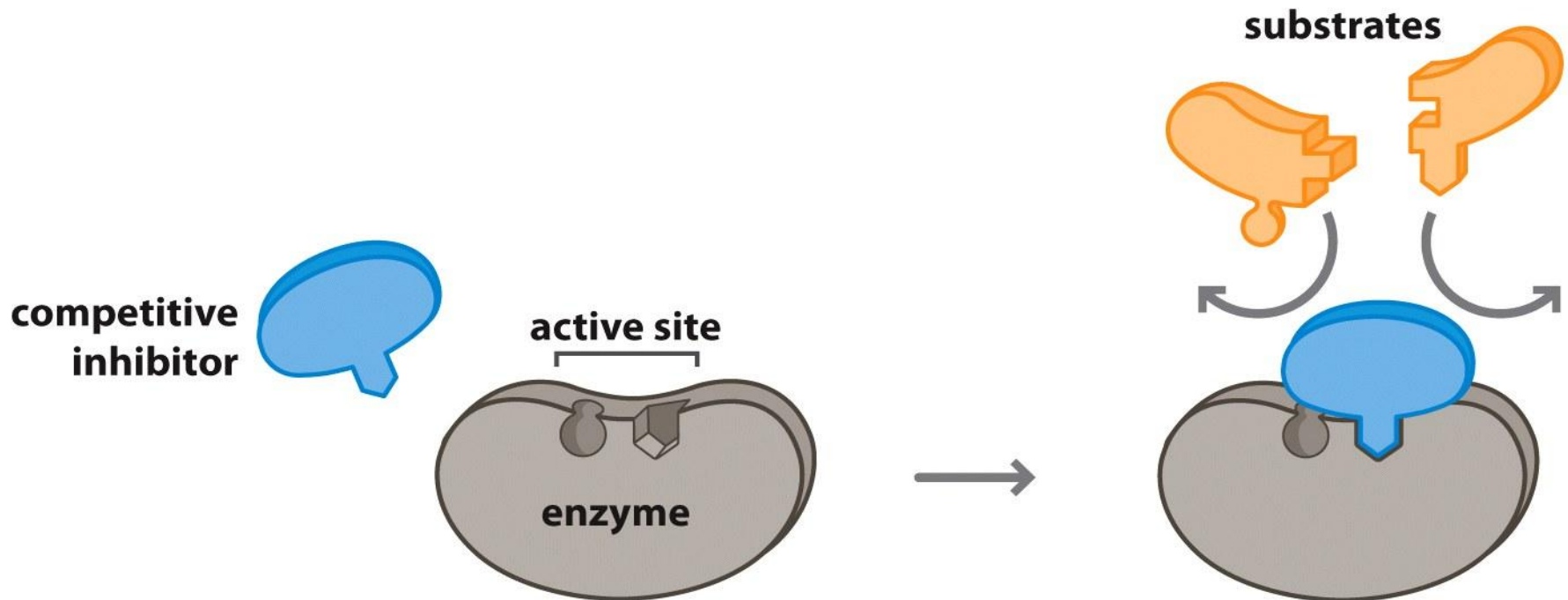
- Two coenzyme carrier molecules frequently encountered in biological reactions are:
 - NAD^+ = nicotinamide adenine dinucleotide
 - FAD = flavin adenine dinucleotide.

ENZYME INHIBITION

- Enzyme inhibition takes place in three ways:
 - Competitive inhibition
 - Allosteric inhibition
 - Feedback inhibition

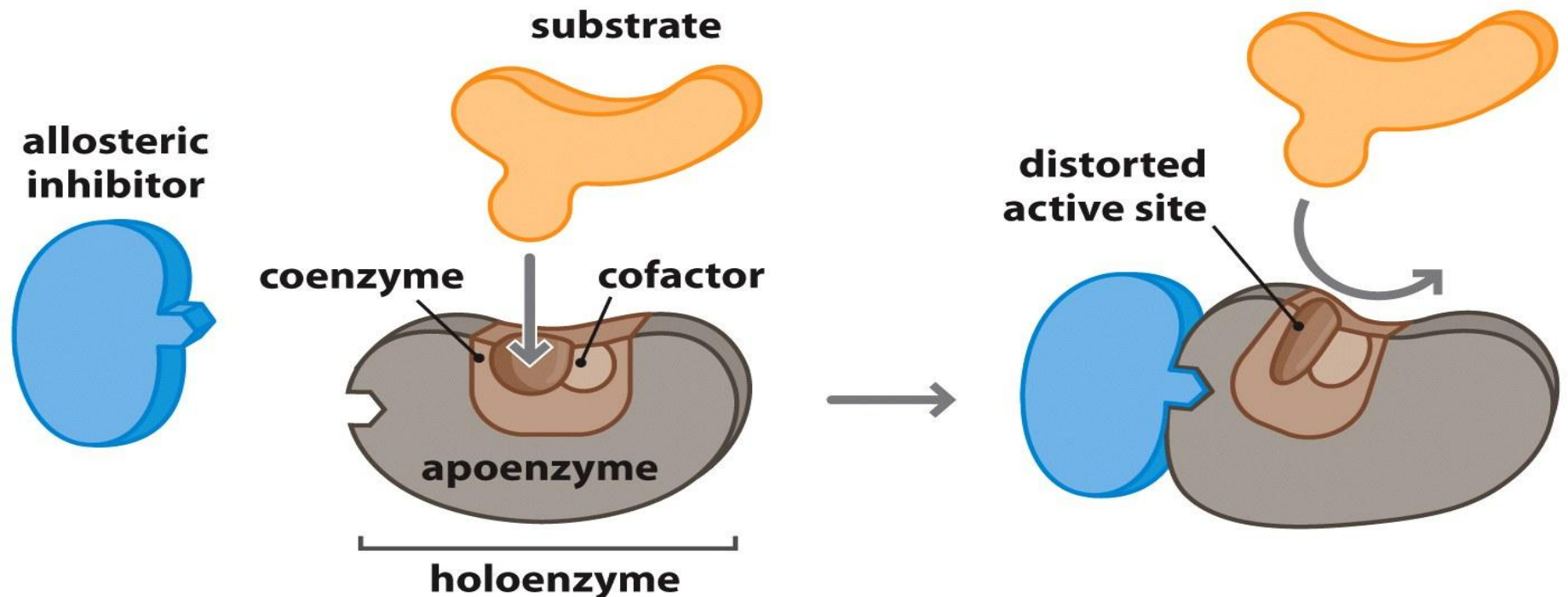
COMPETITIVE INHIBITION

- The **inhibitor** molecule is similar in structure to the substrate and competes with the substrate to bind to the active site.
- When the inhibitor has bound to the active site, the substrate cannot bind.
- The **binding** dependent upon the relative numbers of inhibitor molecules and substrate molecules present.



ALLOSTERIC INHIBITION

- Inhibitor molecules bind to a part of the enzyme away from the active site: the **allosteric site**.
- This binding **changes the shape of the active site** in such a way that it can no longer fit properly with the substrate.



FEEDBACK INHIBITION

- The final product in a pathway accumulates and begins to bind to and inactivate the enzyme
- when the level of end product decreases, the inhibition stops and the enzyme begins to function again.

Continue...

- Example: Inhibitors are often used as drugs, in many cases to prevent detrimental reactions in an organism. Aspirin, inhibits the enzymes that causes pain and inflammation.
- However, inhibitors can also be poisonous. Cyanide is a lethal toxin because it competitively inhibits cytochrome coxidase, an enzyme involved with cellular respiration.

FACTORS THAT AFFECT ENZYME REACTIONS

- Three major factors affect enzyme activity:
 - **Temperature** – Can break hydrogen bonds and change shape
 - **pH** – Can break hydrogen bonds and change shape
 - **Concentration** of substrate, product & enzyme – Lower numbers of substrate, product, and enzyme molecules means a lower level of activity.