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فيزياء المواد Physics of Materials

المرحلة الثالثة الكورس الاول

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10. Thermal properties الخواص الحرارية

السعة الحرارية Heat Capacity

A solid material, when heated عندما تسخن, experiences عندما in temperature رجة الحرارة signifying درجة الحرارة that some energy has been absorbed على signifying هي خاصية that is indicative على that is indicative هي خاصية to absorbe قدرة المادة property المتساص heat from the external surroundings المحيط الخارجي it represents عثر the amount الخارجي of energy required الطاقة المطلوبة to produce الطاقة المطلوبة In mathematical terms, the heat capacity C is expressed as follows:

$$C = \frac{dQ}{dT}$$

where dQ is the energy required to produce a dT temperature change. heat capacity is specified per mole of material (e.g., J/mol. K, or cal/mol .K).

التمدد الحراري Thermal Expansion

Most معظم solid materials expand تتمدد upon heating بالحرارة and contract بالبرودة when cooled بالبرودة The change بالطول with temperature for a solid material may be expressed as follows:

$$\frac{l_f-l_0}{l_0}=\alpha_l(T_f-T_0)$$

 $\frac{\Delta l}{l_0} = \alpha_l \Delta T$

or

where l_0 and l_f represent, respectively, initial بدائي and final نهائي lengths with the temperature change تغير درجة الحرارة from T_0 to T_f . The parameter المعامل الخطي التمدد الحراري; it is a material property خاصية that is indicative تدل على of the extent مدى to which a material expands بالتسخين upon heating ولها وحدات and has units مقلوب temperature ${}^{\circ}C^{-1}$.

heating or cooling affects تؤثر all the dimensions مع of a body, with جميع ابعاد of a body, with تغير ات الحجم volume changes في الحجم with temperature may be computed تحسب from

$$\frac{\Delta V}{V_0} = \alpha_v \Delta T$$

where ΔV and V_0 are the volume change التغير بالحجم and the original volume التغير بالحجم and the original volume المعلى, respectively على التتابع. And α_v symbolizes يرمز الى the volume coefficient of thermal expansion المعامل الحجمي للتمدد الحراري.

التوصيلية الحرارية Thermal Conductivity

Thermal conduction التوصيل الحراري is the phenomenon خاهرة by which heat is transported خاهرة from high حمن العالي to low الواطي temperature regions مناطق that characterizes الحرارة the da substance المحادة of a substance الحرارة the ability مقدرة of a material to transfer الحرارة the the thermal conductivity. It is best defined افضل تعریف in terms of the expression.

$$q = -k \frac{dT}{dx}$$

where q denotes فيض الحرارة, or heat flow جريان الحرارة, per unit time per unit area, \mathbf{k} is the thermal conductivity, and dT/dx is the temperature gradient التدرج الحراري through the conducting medium ...

10.4 Mechanisms of Heat Conduction

Heat is transported بكلا من by both في المادة الصلبة by both موجات اهتزاز الشبيكة by both الالكترونات ophonon الفونون) and free electrons الحرة of these مع كل with each ترافق with each ترافق of these mechanisms, and the total conductivity is associated التوصيلية الكلية is the sum جمع of the two contributions التوزيعين, or

$$k = k_l + k_e$$

where k_l and k_e represent the lattice vibration and electron thermal conductivities, respectively.

The thermal energy المرافقة associated المرافقة with phonons or lattice waves is transported حركتها in the direction باتجاه of their motion حركتها. The k_l contribution باتجاه from a net movement الدركة الصافية of phonons from high- to low temperature regions مناطق of a body across عبر which a temperature gradient الموجود exists التدرج الحراري الحراري in electronic thermal conduction التوصيلية الحرارية الالكترونية.

المعادن 10.4.1 Metals

In high-purity metals, the electron mechanism الية الالكترون of heat transport اكثر كفائة من is much more efficient لانتقال الحرارة than the phonon contribution لانتقال العروب because electrons لان الالكترونات are not as easily scattered مثل as لاتستطار بسهولة and have higher velocities الفونونات Furthermore, وتمتلك سرع عالية extremely good conductors موصلة جيدة of heat الحرارة because relatively نسبيا large numbers المحروبات الحرة of free electrons التوصيل that participate والتي تشارك in thermal conduction الحراري الحراري The thermal conductivities of several of the common metals

are given in Table 10.1; values generally range between about 20 and 400 W/m.K.

10.4.2 Ceramics السيراميك

Non-metallic materials مواد غير معدنية are thermal insulators عوازل حرارية because they lack مواد غير المعدنية large numbers الإلكترونات الحرة of free electrons الالكترونات الحرة the phonons are primarily responsible المسؤول الأول for thermal conduction: k_e is much smaller than k_l .

Thermal conductivity values قيم for a number of ceramic materials are contained نحصل عليها in Table 10.1; room-temperature thermal conductivities range between approximately 2 and 50 W/m. K

البوليمرات Polymers البوليمرات

Thermal conductivities for most polymers are on the order of 0.3 W/m.K. For these materials, energy transfer انتقال الطاقة is accomplished ينجز by the vibration جزيئات السلاسل of the chain molecules الاهتزاز The magnitude قيمة of the thermal conductivity depends درجة on the degree عوازل حرارية Polymers are often used as thermal insulators عوازل حرارية وصيايتها الحرارية القايلة because of their low thermal conductivities

Table 10.1 the thermal properties for a Variety of Materials

Material	$(J/kg \cdot K)^a$	$[({}^{\circ}C)^{-1} \overset{\alpha_I}{\times} 10^{-6}]^b$	$(W/m \cdot K)^c$	$[\Omega \cdot W/(K)^2 \times 10^{\circ}]$
		Metals		
Aluminum	900	23.6	247	2.20
Copper	386	17.0	398	2.25
Gold	128	14.2	315	2.50
Iron	448	11.8	80	2.71
Nickel	443	13.3	90	2.08
Silver	235	19.7	428	2.13
Tungsten	138	4.5	178	3.20
1025 Steel	486	12.0	51.9	_
316 Stainless steel	502	16.0	15.9	_
Brass (70Cu-30Zn)	375	20.0	120	_
Kovar (54Fe-29Ni-17Co)	460	5.1	17	2.80
Invar (64Fe-36Ni)	500	1.6	10	2.75
Super Invar (63Fe-32Ni-5Co)	500	0.72	10	2.68
		Ceramics		
Alumina (Al ₂ O ₃)	775	7.6	39	_
Magnesia (MgO)	940	13.5^{d}	37.7	_
Spinel (MgAl ₂ O ₄)	790	7.6^{d}	15.0^{e}	_
Fused silica (SiO ₂)	740	0.4	1.4	_
Soda-lime glass	840	9.0	1.7	_
Borosilicate (Pyrex) glass	850	3.3	1.4	_
		Polymers		
Polyethylene (high density)	1850	106-198	0.46-0.50	_
Polypropylene	1925	145-180	0.12	_
Polystyrene	1170	90-150	0.13	_
Polytetrafluoroethylene (Teflon)	1050	126–216	0.25	_
Phenol-formaldehyde, phenolic	1590-1760	122	0.15	_
Nylon 6,6	1670	144	0.24	_
Polyisoprene	_	220	0.14	_

10.5 Thermal Stresses الاجهادات الحراري

in a body هي أجهاد are stresses الأجهادات الحرارية induced في as a result of نتيجة الى changes الجسم in temperature في درجة الحرارة. An of thermal stresses is وطبيعة and nature وطبيعة important مهمة because these stresses لأن هذه الأجهادات can lead مهمة to fracture الكسر or undesirable plastic deformation الكسر or undesirable plastic deformation . تشوه البلاستيك غير مرغوب فيه

a homogeneous متناظر and isotropic متباطر solid rod اسطوان صلبة uniformly او تبرد that is heated بانتظام uniformly او تبرد التمدد For free expansion مسلط are imposed لايوجد تدرج حراري For free expansion or contraction الحر, the rod will be stress free.

The magnitude of the stress σ resulting from a temperature change from T_0 to T_f is

$$\sigma = E\alpha_l(T_0 - T_f) = E\alpha_l\Delta T$$

where E is the modulus of elasticity and α_l is the linear coefficient of thermal expansion التمدد الحراري.

Example 10.1 Thermal Stress Created upon Heating

A brass rod is to be used in an application requiring its ends to be held rigid. If the rod is stress free at room temperature [20 °C], what is the maximum temperature to which the rod may be heated without exceeding a compressive stress of 172 MPa, Assume a modulus of elasticity of 100 GPa for brass. the magnitude of the linear coefficient of thermal expansion is 20.0 x 10⁻⁶ (°C)⁻¹.

Solution

To solve this problem, where the stress of 172 MPa is taken to be negative. Also, the initial temperature T_0 is 20 °C, and the magnitude of the linear coefficient of thermal expansion is 20.0 x 10^{-6} (°C)⁻¹. Thus, solving for the final temperature T_f yields

$$\sigma = E\alpha_l(T_0 - T_f)$$

$$T_f = T_0 - \frac{\sigma}{E\alpha_l}$$

$$= 20^{\circ}\text{C} - \frac{-172 \text{ MPa}}{(100 \times 10^3 \text{ MPa})[20 \times 10^{-6} \text{ (°C)}^{-1}]}$$

$$= 20^{\circ}\text{C} + 86^{\circ}\text{C} = 106^{\circ}\text{C}$$

Reference

1- Materials _Science_ and _Engineering_9th .pdf · version 1