جامعة الانبار

كلية: الصيدلة

قسم: فرع العلوم المحتبرية السريرية

اسم المادة باللغة العربية: الكيمياء الحياتية

اسم المادة باللغة الإنكليزية: Biochemistry I

المرحلة: الثالثة

التدريسي: المدرس المساعد زهير عبدالستار احمد

عنوان المحاضرة باللغة العربية: تجربة تفاعل الننهدرين

عنوان المحاضرة باللغة الإنكليزية: Ninhydrin reaction experiment

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### College of Pharmacy third level

practical lab. (3) (proteins)



### 2- Ninhydrin reaction:

This reaction is used to detect the free amino group in peptide compounds and amino acids.

It is a chemical powder which is soluble in ethanol or acetone at room temperature. When a ninhydrin solution is sprayed over fingerprints (typically using a simple spray bottle), the ninhydrin reacts with the amino acids in the fingerprint residue.



## Ninhydrin reaction:

### **Principle:**

• When a solution of ninhydrin is heated with the amino acid, which is considered a reducing agent, the amino acid is oxidized by removing the amino group from it by means of hydrindantine, and the ammonia is released and condensed with hydrindantine, forming a purple compound.

#### Ninhydrin reaction •

This test is a chemical test that is used to check whether a given analyte contains amines or  $\alpha$ amino acids. In this test, ninhydrin (a chemical compound with the formula ( $C_0H_6O_4$ ) name: 2,2dihydroxyindane-1,3-dione) is added to a test solution of the analyte. The development of a deep blue color indicates the presence of ammonia, acids in the .primary/secondary amines, or amino analyte



### Ninhydrin reaction:

The chemical reaction that generally occurs in a positive ninhydrin test is illustrated below.

OH 
$$\frac{NH_2CHCOOH}{HO}$$
  $+$   $\frac{O}{C}$   $+$   $CO_2$   $+$   $4H_2O$ 

Ninhydrin Test



## Ninhydrin reaction:

#### **Method:**

- put 1 ml of tryptophan amino acid into a test tube+ 2 drops of a 0.2% ninhydrin solution.
- •heat the mixture in a boiling Water Bath for (2-5min).
- A violet color is produced.



#### **Ninhydrin reaction**



### Ninhydrin Test Result Interpretation

- 1. For ammonia, <u>primary/secondary amines</u>, and amino acids, deep <u>purple color</u> is obtained.
- 2. For hydroxyproline and proline, a yellow color is obtained.
- 3. For asparagine, brown color is obtained.

